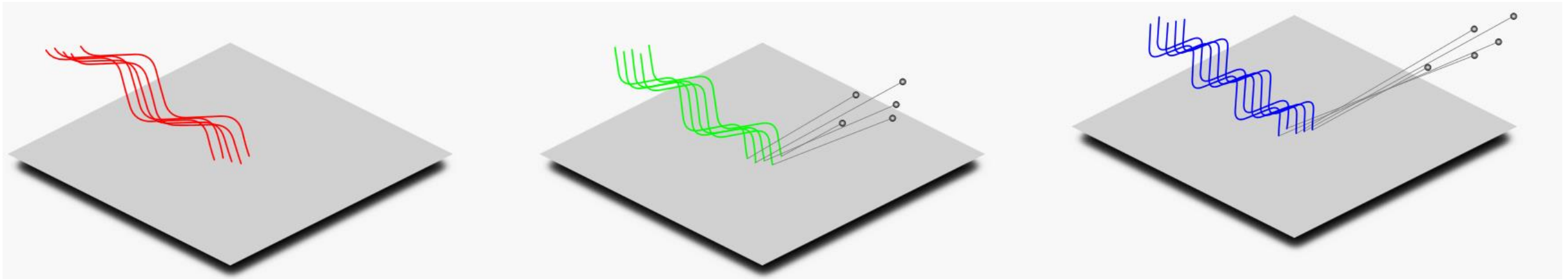
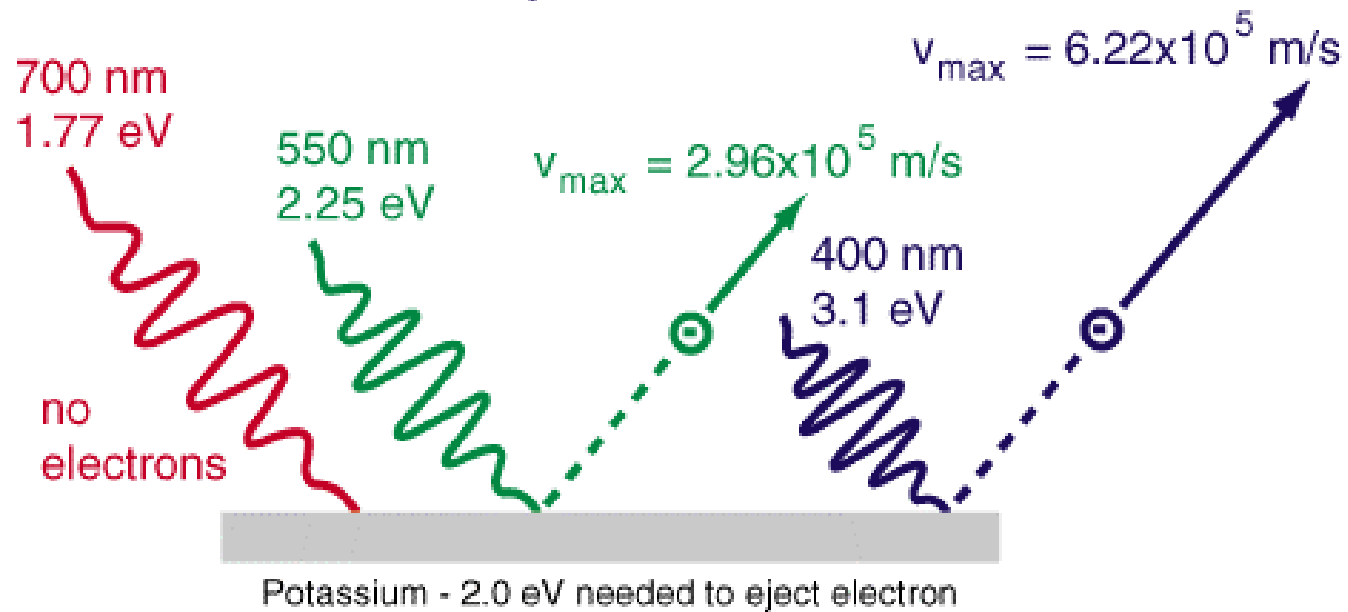


Effet photoélectrique



$$E_{\text{photon}} = h\nu$$



$$1\text{eV} = 1,602 \times 10^{-19}\text{J}$$

$$1\text{eV} \sim 100\text{kJ}\cdot\text{mol}^{-1}$$

Photoelectric effect

$$K_{\text{e}} = h\nu - \Phi$$

$$\frac{1}{2}mv_{\text{e}}^2 = h\nu - \Phi$$

